
(1)

(2)

(3)

(4)

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# Methyl trans-2-(Diphenylmethyl)-5-(hydroxymethyl)-3-isoxazolidinylacetate and cis-5-(Ethoxymethyl)-2-phenyl-3-isoxazolidinyl Phenyl Ketone 

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#### Abstract

C}_{19} \mathrm{H}_{21} \mathrm{NO}_{4}(1): M_{r}=327 \cdot 38\), monoclinic, $P 2_{\mathrm{I}} / n, a=10.103$ (3), $b=20.253$ (4), $c=9.273$ (2) $\AA$, $\beta=115.49(2)^{\circ}, \quad V=1713(1) \AA^{3}, \quad Z=4, \quad D_{m}=$ 1.25 (2), $D_{x}=1.27 \mathrm{~g} \mathrm{~cm}^{-3}, \lambda($ Mo K $\alpha$ ) $=0.71069 \AA$, $\mu=0.52 \mathrm{~cm}^{-1}, F(000)=696.0$. Final conventional $R$ $=0.046$ for 2742 unique observed reflections. $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{NO}_{3}$ (2): $M_{r}=311 \cdot 37$, orthorhombic, Pbca, a $=18.213$ (3),$\quad b=20.451$ (4), $c=8.904$ (2) $\AA, \quad V=$ 3317 (3) $\AA^{3}, Z=8, D_{m}=1.24$ (2), $D_{x}=1.25 \mathrm{~g} \mathrm{~cm}^{-3}$, $\lambda($ Mo $K \alpha)=0.71069 \AA, \quad \mu=0.48 \mathrm{~cm}^{-1}, \quad F(000)=$ 1328.0. Final conventional $R=0.066$ for 1836 unique observed reflections. The stereochemistry of both compounds was established unequivocally. The conformation of the isoxazolidine ring of (1) has approximate $C_{2}$ symmetry through $\mathrm{C}(5)$, while that


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of (2) has $C_{s}$ symmetry, $\mathrm{N}(2)$ being the out-of-plane atom. Two substituents at $\mathrm{N}(2)$ and $\mathrm{C}(3)$ of both compounds have a trans-diaxial orientation in the crystal structure.

Introduction. The investigation of regio- and stereoselectivities in the 1,3-dipolar cycloaddition of nitrones is one of the most stimulating problems from both theoretical and synthetic points of view (Padwa, 1984). In order to clarify the selectivities, it is essential to elucidate the stereochemistry of the isoxazolidine ring. However, it is sometimes difficult to assign the stereochemistry by ordinary ${ }^{1} \mathrm{H}$ NMR techniques because the isoxazolidine ring may exist in many different conformers.

Table 1. Final atomic coordinates of the non-H atoms $\left(\times 10^{4}\right)$ and equivalent isotropic temperature factors with e.s.d.'s in parentheses

|  | $x$ | $y$ | $z$ | $B_{\text {eq }}{ }^{*}\left(\AA^{2}\right)$ |
| :---: | :---: | :---: | :---: | :---: |
| (1) |  |  |  |  |
| $\mathrm{O}(1)$ | 2216 (2) | 657 (1) | 2501 (2) | $4 \cdot 3$ (1) |
| N(2) | 1467 (2) | 1089 (1) | 3112 (3) | $3 \cdot 8$ (1) |
| C(3) | 2583 (3) | 1198 (1) | 4822 (3) | $4 \cdot 0$ (1) |
| C(4) | 4042 (3) | 1222 (2) | 4798 (4) | $4 \cdot 3$ (1) |
| C(5) | 3787 (3) | 770 (2) | 3389 (4) | $4 \cdot 2$ (1) |
| C(6) | 2521 (3) | 631 (2) | 5840 (4) | $4 \cdot 8$ (1) |
| O(7) | 3355 (3) | 184 (1) | 6298 (4) | 10.7 (2) |
| $\mathrm{O}(8)$ | 1456 (3) | 693 (1) | 6199 (3) | $7 \cdot 2$ (1) |
| C(9) | 1313 (5) | 180 (2) | 7210 (5) | $9 \cdot 1$ (2) |
| C(10) | 4295 (4) | 1044 (2) | 2220 (4) | 6.0 (2) |
| $\mathrm{O}(11 A)$ | 5746 (4) | 1116 (2) | 2892 (5) | $5 \cdot 8$ (2) |
| $\mathrm{O}(11 B)$ | 4334 (8) | 606 (3) | 1197 (7) | $8 \cdot 1$ (3) |
| C(12) | 1099 (3) | 1710 (1) | 2153 (3) | $3 \cdot 5$ (1) |
| C(13) | 217 (3) | 1576 (1) | 331 (3) | $3 \cdot 5$ (1) |
| C(14) | 302 (3) | 2022 (2) | -742 (4) | $4 \cdot 3$ (1) |
| C(15) | - 530 (4) | 1947 (2) | - 2408 (4) | $5 \cdot 3$ (2) |
| C(16) | - 1460 (4) | 1421 (2) | - 3036 (4) | 5.4 (2) |
| C(17) | -1550 (4) | 967 (2) | -1978 (4) | $5 \cdot 0$ (1) |
| C(18) | -715 (3) | 1043 (2) | -307 (4) | $4 \cdot 3$ (1) |
| C(19) | 258 (3) | 2151 (1) | 2735 (3) | $3 \cdot 5$ (1) |
| C(20) | 803 (4) | 2759 (2) | 3411 (4) | $4 \cdot 5$ (1) |
| C(21) | 1 (4) | 3173 (2) | 3889 (4) | 5.4 (2) |
| C(22) | -1345 (4) | 2973 (2) | 3676 (4) | $5 \cdot 6$ (2) |
| C(23) | - 1887 (4) | 2366 (2) | 3026 (4) | $5 \cdot 7$ (2) |
| C(24) | -1092 (3) | 1954 (2) | 2557 (4) | $4 \cdot 8$ (1) |
| (2) |  |  |  |  |
| $\mathrm{O}(1)$ | 7052 (2) | 4351 (2) | -400 (4) | $4 \cdot 0$ (2) |
| N(2) | 7533 (3) | 4814 (2) | 294 (5) | $3 \cdot 5$ (2) |
| C(3) | 7092 (3) | 5058 (3) | 1562 (6) | $3 \cdot 5$ (3) |
| C(4) | 6680 (4) | 4466 (3) | 2139 (7) | $4 \cdot 1$ (3) |
| C(5) | 6662 (3) | 4010 (3) | 787 (7) | 4.0 (3) |
| C(6) | 6593 (3) | 5590 (3) | 913 (6) | $3 \cdot 9$ (3) |
| O(7) | 5935 (2) | 5513 (2) | 861 (6) | $5 \cdot 8$ (2) |
| C(8) | 6945 (3) | 6187 (3) | 283 (6) | $3 \cdot 7$ (3) |
| C(9) | 6610 (4) | 6497 (3) | -931 (7) | $4 \cdot 7$ (3) |
| C(10) | 6931 (4) | 7048 (3) | - 1546 (8) | $5 \cdot 5$ (4) |
| C(11) | 7572 (4) | 7296 (3) | -962 (8) | $5 \cdot 1$ (3) |
| C(12) | 7893 (4) | 7001 (3) | 236 (8) | $5 \cdot 1$ (3) |
| C(13) | 7590 (4) | 6443 (3) | 847 (7) | 4.4 (3) |
| C(14) | 5916 (4) | 3857 (3) | 202 (8) | $5 \cdot 1$ (4) |
| O(15) | 5971 (3) | 3390 (2) | -941 (6) | 7.7 (3) |
| C(16) | 5278 (5) | 3254 (4) | 1501 (12) | 12.5 (6) |
| C(17) | 5263 (5) | 2581 (4) | -2271 (11) | 11.4 (6) |
| C(18) | 8240 (3) | 4545 (3) | 587 (6) | $3 \cdot 5$ (3) |
| C(19) | 8739 (3) | 4897 (3) | 1443 (7) | $4 \cdot 6$ (3) |
| C(20) | 9451 (4) | 4675 (4) | 1606 (9) | $5 \cdot 5$ (4) |
| C(21) | 9668 (4) | 4102 (4) | 935 (9) | $5 \cdot 7$ (4) |
| C(22) | 9176 (4) | 3750 (4) | 114 (8) | $5 \cdot 7$ (4) |
| C(23) | 8463 (4) | 3965 (3) | -74 (7) | 4.6 (3) |

* $B_{\text {eq }}$ is the arithmetic mean of the principal axes of the thermal ellipsoid.

In the course of our work on 1,3-dipolar cycloaddition, we obtained several isoxazolidine derivatives, some of which were obtained in crystalline form. The reaction of $N$-diphenymethyl $\alpha$-methoxycarbonylmethanimine $N$-oxide with acryl aldehyde gave a mixture of two epimeric adducts in the ratio $82: 18$. The aldehyde group was easily reduced by sodium borohydride at low temperature to give crystalline alcohols [(1) and (1a)], from which the major isomer (1) was separated by recrystallization. The stereochemistry of (1) has previously been assumed to be $3,5-$ trans by converting it to allohydroxyproline (Hara, Inouye \& Kakisawa, 1981).

The reaction of $N$-phenyl- $\alpha$-benzoylmethanimine $N$-oxide with 3 -ethoxypropene gave a mixture of regioisomers (2) and (2a) [(2):(2a)=91•9]. The stereochemistry of the major isomer (2) remained

Table 2. Selected bond distances ( $\AA$ ), bond angles $\left({ }^{\circ}\right)$ and torsion angles ( ${ }^{\circ}$ ) around the isoxazolidine ring with e.s.d.'s in parentheses

|  | (1) | (2) |
| :---: | :---: | :---: |
| $\mathrm{O}(1)-\mathrm{N}(2)$ | 1.424 (3) | 1.431 (5) |
| $\mathrm{N}(2)-\mathrm{C}(3)$ | 1.515 (3) | 1.472 (7) |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | 1.484 (4) | 1.514 (8) |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | 1.525 (4) | 1.524 (8) |
| $\mathrm{C}(5)-\mathrm{O}(1)$ | 1.458 (3) | 1.451 (7) |
| N(2)-C(subst) | 1.493 (3) | 1.425 (7) |
| C(3)-C(subst) | 1.506 (4) | 1.530 (8) |
| C(5)-C(subst) | 1.491 (4) | 1.489 (8) |
| $\mathrm{C}(3)-\mathrm{H}(3)$ | 0.95 (3) | 1.00 (5) |
| $\mathrm{C}(4)-\mathrm{H}(4 A)$ | 1.00 (2) | 0.98 (6) |
| $\mathrm{C}(4)-\mathrm{H}(4 B)$ | 0.98 (3) | 1.00 (6) |
| $\mathrm{C}(5)-\mathrm{H}(5)$ | 1.03 (3) | 0.94 (5) |
|  | (1) | (2) |
| $\mathrm{N}(2)-\mathrm{O}(1)-\mathrm{C}(5)$ | 108.3 (2) | 107.6 (4) |
| $\mathrm{O}(1)-\mathrm{N}(2)-\mathrm{C}(3)$ | 102.8 (2) | 102.9 (4) |
| $\mathrm{O}(1)-\mathrm{N}(2)-\mathrm{C}($ subst) | 108.6 (2) | 112.1 (4) |
| $\mathrm{C}(3)-\mathrm{N}(2)-\mathrm{C}$ (subst) | 112.5 (2) | 118.9 (4) |
| $\mathrm{N}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | $106 \cdot 6$ (2) | 105.0 (4) |
| $\mathrm{N}(2)-\mathrm{C}(3)-\mathrm{C}($ subst) | 109.4 (2) | $105 \cdot 9$ (4) |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}($ subst) | 110.1 (2) | 113.7 (5) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 101.0 (2) | 103.4 (5) |
| $\mathrm{O}(1)-\mathrm{C}(5)-\mathrm{C}(4)$ | 108.6 (2) | 105.7 (4) |
| $\mathrm{O}(1)-\mathrm{C}(5)-\mathrm{C}($ subst) | 106.2 (2) | 107.0 (5) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}$ (subst) | 114.5 (3) | $115 \cdot 1$ (5) |

Torsion angles and the observed $J(\mathrm{~Hz})$; e.s.d's in angles are 2-5

|  | $(1)$ |  | (2) |  |
| :--- | ---: | ---: | ---: | ---: |
|  | Angle | $J$ | Angle | $J$ |
| $\mathrm{H}(3)-\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{H}(4 A)$ | 31.5 | 8.5 | 18.9 | 8.5 |
| $\mathrm{H}(3)-\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{H}(4 B)$ | 90.5 | 4.2 | 99.6 | 4.2 |
| $\mathrm{H}(4 A)-\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{H}(5)$ | 140.0 | 7.9 | 5.7 | 8.3 |
| $\mathrm{H}(4 B)-\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{H}(5)$ | 16.4 | 7.6 | 127.8 | 7.2 |

uncertain (Niwayama, Dan, Inouye \& Kakisawa, 1985).

We present here a single-crystal X-ray analysis of two isoxazolidine derivatives (1) and (2) in order to elucidate the ambiguous stereochemistry of those adducts.

Experimental. Isoxazolidine (1): colorless prisms from benzene; $D_{m}$ by flotation in aqueous KI ; systematic absences (from Weissenberg photographs) $h 00$ for $h=2 n, 0 k 0$ for $k=2 n, 00 l$ for $l=2 n$ and ( $h 0 l$ ) for $h+l=2 n$; crystal $0.38 \times 0.27 \times 0.28 \mathrm{~mm}$; Rigaku AFC-5 automated four-circle diffractometer; graphite-monochromated Mo $K \alpha$ radiation, 50 kV and 160 mA ; unit-cell dimensions by least-squares refinement from 25 reflections with $19<2 \theta<21^{\circ}$; $2 \theta-\omega$ scan, scan rate $2^{\circ} \mathrm{min}^{-1}, \quad \Delta \omega=(1.0+$ $0.5 \tan \theta)^{\circ}$; three standard reflections measured every 50 reflections with $\pm 1 \%$ fluctuation; 3039 unique reflections, $2 \theta_{\text {max }}=50^{\circ}(-12 \leq h \leq 10,0 \leq k \leq 24,0$ $\leq l \leq 11) ; 2742$ reflections with $\left|F_{o}\right|>3 \sigma\left(F_{o}\right)$ used for structure determination; intensities corrected for Lorentz and polarization, not for absorption; structure solved by direct methods (SHELX76; Sheldrick, 1976); first $E$ map ( $E \geq 1.5$ ) revealed positions of all non-H atoms; disorder at the hydroxymethyl group $[\mathrm{C}(10)-\mathrm{O}(11)]$, site occupancy $0 \cdot 6: 0 \cdot 4$; all H atoms except those around disordered part and of methyl group located from difference

Fourier synthesis; H atoms on methyl group given fixed geometries $\mathrm{C}-\mathrm{H} 1.04 \AA$ and $U 0.1 \AA^{2}$; final refinement on $F$ with anisotropic temperature factors for all non- H atoms and with isotropic H atoms (except H atoms mentioned above), $R=0.046, w R=$ $0.046, w=1 / \sigma^{2}\left(F_{o}\right)$ from counting statistics, $S=$ 0.77 ; $(\Delta / \sigma)_{\max }<0.2 ;|\Delta \rho|_{\max }$ in final difference Fourier map $=0.19 \mathrm{e}^{-3} \AA^{-3}$.

Isoxazolidine (2): colorless prisms from hexanebenzene; $D_{m}$ by flotation in aqueous KI; systematic absences (from Weissenberg photographs) $h k 0$ for $h$ $=2 n, 0 k l$ for $k=2 n$ and $h 0 l$ for $l=2 n$; crystal $0 \cdot 30 \times$ $0.25 \times 0.08 \mathrm{~mm}$; collection of the reflection data and structural analyses similar to (1); three standard


Fig. 1. ORTEP (Johnson, 1976) drawings of the molecules with thermal ellipsoids at the $50 \%$ probability level and numbering schemes. H atoms are represented by circles of radius $0 \cdot 1 \AA$. (a) (1), the disordered part is shown by the dotted line. (b) (2).
reflections measured every 56 reflections with $\pm 3 \%$ fluctuation; 2176 unique reflections, $2 \theta_{\max }=45^{\circ}(0 \leq$ $h \leq 19,0 \leq k \leq 22,0 \leq l \leq 9)$; 1838 reflections with $\left|F_{o}\right|>3 \sigma\left(F_{o}\right)$ used for structure determination; two large reflections apparently seriously affected by extinction $(200,020)$ omitted from final refinement; first $E$ map $(E \geq 1 \cdot 5)$ revealed positions of 20 non-H atoms; successive difference Fourier syntheses determined positions of all non-H atoms; all H atoms located from difference Fourier synthesis; H atoms on methyl group given fixed geometries $\mathbf{C}-\mathrm{H}$ $1.04 \AA$ and $U 0.1 \AA^{2}$; bond length $C(16)-C(17)$ constrained to $1.54 \AA$ with $\sigma=0.002$; final refinement on $F$ with anisotropic temperature factors for all non- H atoms and with isotropic H atoms (except H atoms mentioned above), $R=0.066, w R=$ $0.066, S=2.17 ; \quad(\Delta / \sigma)_{\text {max }}<0.2 ; \quad|\Delta \rho|_{\text {max }}$ in final difference Fourier map $=0.13 \mathrm{e} \AA^{-3}$. All numerical calculations carried out on a Facom M-382 computer in the Science Information Processing Center of the University of Tsukuba; scattering factors from International Tables for X-ray Crystallography (1974).


Fig. 2. The crystal packings. (a) (1); (b) (2).

Discussion. The final atomic parameters are listed in Table 1.* Molecular structures of (1) and (2) are shown in Fig. 1 with numbering schemes. The stereochemistry of (1) is shown to be 3,5-trans as previously thought and that of (2) as 3,5 -cis unequivocally. The geometries of the isoxazolidine rings in both compounds are similar but a slight difference is observed in some of the bond lengths and angles as shown in Table 2. The conformation of the isoxazolidine ring in the crystal has approximate $C_{2}$ symmetry through $\mathrm{C}(5)$ in (1) [deviation from $\mathrm{O}(1)-\mathrm{C}(5)-\mathrm{C}(4)$ plane: $\mathrm{N}(2)-0.21 \AA$ and $\mathrm{C}(3)$ $+0.34 \AA$ ], and $C_{s}$ symmetry [with $\mathrm{N}(2)$ the out-ofplane atom] in (2) $[\mathrm{N}(2)-0.56 \AA]$. In both compounds, the configuration of the substituents at the $\mathrm{N}(2)$ and $\mathrm{C}(3)$ positions is trans-diaxial due to the steric repulsion of the two substituents. The torsion angles between the protons on the ring are show in Table 2 together with the coupling constants observed in $\mathrm{CDCl}_{3}$. The angles and the coupling constants are not necessarily correlated; this is apparently due either to the pseudorotation or to equilibrium among several conformers in solution.

* Lists of structure factors, anisotropic thermal parameters, H -atom coordinates, and bond lengths have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 52448 ( 22 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

The crystal packings are shown in Fig. 2. In (1), both the disordered hydroxyl groups $[\mathrm{C}(10)-\mathrm{O}(11 A)$ and $\mathrm{C}(10)-\mathrm{O}(11 B)]$ are bonded to the carbonyl O atom $[\mathrm{O}(7)]$ of the ester group $[\mathrm{O}(11 A)(x, y, z) \cdots \mathrm{O}(7)$ $(1-x,-y, 1-z)=2.781(5) \AA, \quad \mathrm{C}(10)-\mathrm{O}(11 A) \cdots$ $\mathrm{O}(7)=100 \cdot 9(3)^{\circ} ; \mathrm{O}(11 B)(x, y, z) \cdots \mathbf{O}(7)(1-x,-y, 1-$ $\left.z)=2 \cdot 956(8) \AA . \quad \mathrm{C}(10)-\mathrm{O}(11 A) \cdots \mathrm{O}(7)=93 \cdot 3^{(4)}{ }^{\circ}\right]$. Other interatomic contacts are within van der Waals distances; the shortest distance is $3.302(5) \AA$ for $\mathrm{O}(1)(x, y, z) \cdots \mathrm{C}(17)(-x,-y,-z)$. On the other hand, only van der Waals contacts are recognized in (2). The shortest distance is 3.348 (7) $\AA$ for $\mathrm{C}(3)$ $(x, y, z) \cdots \mathrm{O}(1)(1 \cdot 5-x, 1-y, 0 \cdot 5+z)$. The somewhat large thermal vibrations observed at $C(16)$ and $C(17)$ in (2) may be due to the presence of some disorder around the ethoxymethyl group.

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# X-ray Structure of the Monoclinic Form of 1,8,15,22-Tetrathia[1.1.1.1]metacyclophane 

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#### Abstract

Tetrathiapentacyclo[19.3.1.1 ${ }^{3,7}$.$1^{9,13} 1^{15,19}$ ]octacosa-1(25),3,5,7(28),9,11,13(27), 15,17,19(26), 21, 23-dodecaene, $\quad \mathrm{C}_{24} \mathrm{H}_{16} \mathrm{~S}_{4}, \quad M_{r}=432 \cdot 6$, monoclinic, $P 2_{1} / n, a=15.418$ (2), $b=7.776$ (1), $c=$ $17 \cdot 048$ (1) $\AA, \beta=93 \cdot 24(1)^{\circ}, V=2040 \cdot 5$ (3) $\AA^{3}, Z=$ $4, D_{x}=1.408 \mathrm{Mg} \mathrm{m}^{-3}, \lambda($ Мо $K \alpha)=0.71069 \AA, \mu=$ $0.46 \mathrm{~mm}^{-1}, F(000)=896, T=293 \mathrm{~K}, R=0.052$ for 1848 observed reflections. The molecule has distorted mm 2 symmetry and a nonplanar conformation, close to that observed in the triclinic form. Two of the


four benzene rings have rather short contacts of $-3.50 \AA$ with the corresponding and almost parallel rings of the neighbouring molecules, suggesting a weak intermolecular stacking interaction.

Introduction. The conformational flexibility of phenylenesulfides resulting in the existence of different conformers in crystals is of significant importance for their possible application as complexones (Ovchinnikov, Zamaev, Shklover, Palyulin, © 1990 International Union of Crystallography

